Model Question Paper

p - Block Elements - Part I

12th Standard

	Chemistry	Reg.No.:			
I Answer all the questions					

I.Answer all the questions
II.Use Blue pen only.

1) An element which was burnt in limited supply of air to give oxide **A** which on treatment with water gives an acid. **B.** Acid **B** on heating gives acid **C** which gives yellow precipitate with $AgNO_3$ solution **A** is

(a) SO_2 (b) NO_2 (c) P_2O_3 (d) SO_3

2) The compound with garlic odour is

(a) P_2O_3 (b) P_2O_5 (c) H_3PO_3 (d) H_3PO_4

3) The shape of PCI_5 is

(a) pyramidal (b) trigonal bipyramidal (c) linear (d) tetrahedral

4) The compound used as smoke screen

(a) PCI_3 (b) PCI_5 (c) PH_3 (d) H_3PO_3

5) Which shows only -1 oxidation state?

(a) fluorine (b) bromine (c) chlorine (d) iodine

Section-B 5 x 3 = 15

6) Write a note on plumbo solvency. (or) Why are lead pipes not used to carry drinking water?

- 7) H_3PO_3 is diprotic. Why?
- 8) Draw the structure of a) PCl_5 b) H_3PO_3
- 9) Prove that P_2O_5 is a powerful dehydrating agent.
- 10) Why H_2O is a liquid while H_2S is a gas. Why?

Section-C 5 x 5 = 25

- 11) What are silicones? How are they prepared? Mention its uses?
- 12) Give an account of manufacture of lead. Describe its important properties and uses.
- 13) How are the following compounds prepared from phosphorus?a) P_2O_3 b) P_2O_5 c) PCl_3 d) PCl_5 e) PH_3
- 14) How fluorine is isolated from their fluorides?Describe its important properties and uses.
- 15) Give a detailed account of the interhalogen compounds with special reference to the compounds involving iodine. Draw their structures.
