Model Question Paper

Ethers - Part I

12th Standard

Chemistry	Reg.No.:
I.Answer all the questions.	
II.Use blue pen only.	
Time : 01:30:00 Hrs	Total Marks : 70
Part-A	5 x 1 = 5
1) The isomerism exhibited by $C_2H_5OC_2H_5$ and is $CH_3-O-CH-CH_3$	
CH_3	
(a) Functional (b) metamerism (c) position (d) chain	
2) Which one of the following is simple ether?	
(a) CH_3 -O- C_2H_5 (b) C_2H_5 -O- CH_3 (c) C_2H_5 -O- C_2H_5 (d) C_3H_7 -O- C_2H_5	
3) Diethyl ether can be decomposed with	
(a) HI (b) KMnO ₄ (c) NaOH (d) H ₂ O	
4) Oxygen atom of ether is	
(a) Very active (b) Replacable (c) Oxidising (d) Comparatively inert	
5) According to Lewis concept of acids and bases, ethers are	
(a) Neutral (b) Acidic (c) Basic (d) Amphoteric	
Part-B	5 x 3 = 15
6) Write the IUPAC names of (a) $C_2H_5OCH_3$ (b) $C_6H_5OC_2H_5$	
7) Ethers should not be heated to dryness. Why?	
8) Ethers are the best solvents of Grignard reagents- Account for this statement.	
9) Write a note on williamson synthesis.	
10) Complete and balance $C_2H_5OH \stackrel{250^{\circ}C}{\xrightarrow{Al_2O_3}}$?	
Part-C	6 x 5 = 30
11) How does diethyl ether react with the following reagents? (a) O_2 / long contact (b) HI in excess (c) PCI ₅ (d) dilute H_2SO_4	
12) Give any two chemical differences betwee <mark>n anisol</mark> e and an alipha <mark>tic ether.</mark>	
13) Write all possible isomers with molecular formula $C_4 H_{10} O$ and name them $M_{10} O$ and $M_{10} O$ a	
14) Explain isomerism in ethers.	
15) Write any three methods of preparation of diethyl ether.	
16) How does diethyl ether react with i. Cl ₂ dark and sunlight ii. O ₂ iii. Con.HCl and Con.H ₂ SO ₄ iv. Lewis acid (BF ₃ and AlCl ₃)	
Part-D	2X10=20
17) a) How does ethers react with HI? Write its significance	
b) Give three methods of preparing anisole from phenol.	
18) a) i) How is anisole manufactured from dimethyl sulphate? ii) What is the action of anisole with BF_3 and $AlCl_3$?	
b) Explain the electrophilic substitution reaction of anisole.	
