

Model Question Paper
Carboxylic Acids - Part V
12th Standard

Chemistry

Reg.No. :

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- I. Answer all the questions.
II. Use blue pen only.
III. Question number 16 is compulsory.

Time : 01:30:00 Hrs

Total Marks : 80

5 x 1 = 5

Part-A

- 1)is used as an antiseptic and astringent
(a) silver nitrate (b) silver lactate (c) silver acetate (d) lactic acid
- 2) Oxalic acid can be prepared from
(a) glycol (b) cyanogen (c) both 'a' and 'b' (d) lactic acid
- 3)is used as an urinary antiseptic
(a) Urotropine (b) C_6H_5COOH (c) CH_3COOH (d) both 'a' and 'b'
- 4) The carbonyl compounds are
(a) aldehydes (b) ketones (c) acids (d) all the above
- 5) $CH_3CO-NH_2 \xrightarrow{P_2O_5} CH_3-C \equiv N$, This reaction is
(a) hydration (b) dehydration (c) decarboxylation (d) deoxidation

Part-B

- 6) Give the structure and IUPAC names of
(i) oxalic acid (ii) malonic acid (iii) succinic acid (iv) adipic acids
- 7) Give the uses of succinic acid.
- 8) How do the groups Cl, OH, NO_2 and carbonyl influence the acidic nature in aromatic acids.
- 9) How does acetic anhydride react with PCl_5 ?
- 10) What happens when acetamide is hydrolysed?

5 x 3 = 15

Part-C

- 11) How does formic acid react with
(i) Na_2CO_3 (ii) Zn, Mg (iii) CH_3OH/H^+
(iv) conc. H_2SO_4 (v) PCl_5
- 12) Write a note on reducing and decarboxylating properties of formic acid
- 13) How will you get oxalic acid from
(i) sodium formate (ii) glycol?
- 14) How do oxalic acid and succinic acid react with the following
(i) NaOH (ii) NH_3 (iii) PCl_5 (iv) on heating (v) acidified $KMnO_4$

10 x 5 = 50

Part-D

- 15) a) How does benzoic acid react with
(i) NaOH (ii) NH_4OH (iii) C_2H_5OH/H^+ ?
- b) Explain how salicylic acid is obtained by Kolbe reaction with mechanism.
- 16) a) a) How are (i) acetyl chloride and (ii) acetic anhydride prepared?
- b) Give the various methods of preparation of methyl acetate

3 x 10 = 30

(OR)

- b) a) Give the uses of (i) acetyl chloride (ii) acetic anhydride and (iii) methyl acetate
- b) Give the various methods of preparation of acetamide.
