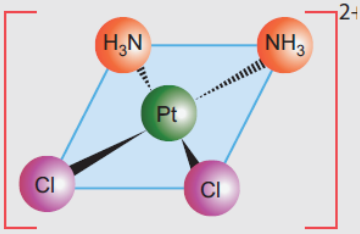
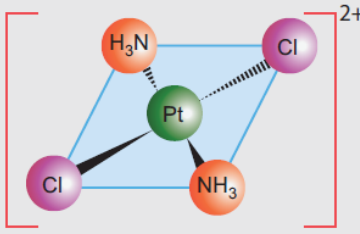
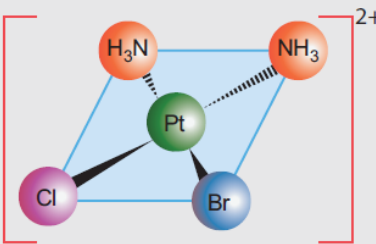
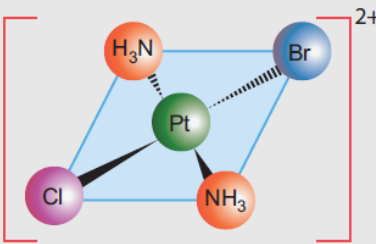
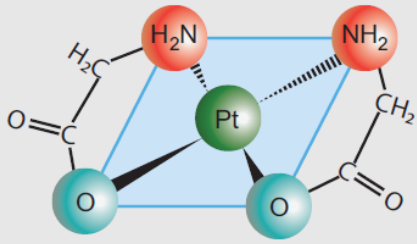
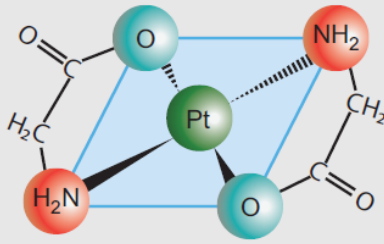


Geometrical isomers:

Geometrical isomerism exists in heteroleptic complexes due to different possible three dimensional spatial arrangements of the ligands around the central metal atom. This type of isomerism exists in square planer and octahedral complexes.

Type	Example	
	Cis Isomer	Trans isomer
MA_2B_2		
MA_2BC		
$M(xy)_2$		
$MABCD$	