Electronic configuration of actinoids

The electronic configuration of actinoids is not definite. The general valence shell electronic configuration of 5f elements is represented as $[Rn]5f^{2-14}6d^{0-2}7s^2$. The following table show the electronic configuration of actinoids.

Name of the element	Atomic number	Symbol	Electronic configuration
Actinium	89	Ac	[Rn] $5f^0$ $6d^1$ $7s^2$
Thorium	90	Th	[Rn] $5f^0$ $6d^2$ $7s^2$
Protactinium	91	Pa	[Rn] $5f^2$ $6d^1$ $7s^2$
Uranium	92	U	[Rn] $5f^3$ $6d^1$ $7s^2$
Neptunium	93	Np	[Rn] $5f^4$ $6d^1$ $7s^2$
Plutonium	94	Pu	[Rn] $5f^6$ $6d^0$ $7s^2$
Americium	95	Am	[Rn] $5f^7$ $6d^0$ $7s^2$
Curium	96	Cm	[Rn] $5f^7$ $6d^1$ $7s^2$
Berkelium	97	Bk	[Rn] $5f^9$ $6d^0$ $7s^2$
Californium	98	Cf	[Rn] $5f^{10}$ $6d^0$ $7s^2$
Einstenium	99	Es	[Rn] $5f^{11}$ $6d^0$ $7s^2$
Fermium	100	Fm	[Rn] $5f^{12}$ $6d^0$ $7s^2$
Mendelevium	101	Md	[Rn] $5f^{13}$ $6d^0$ $7s^2$
Nobelium	102	No	[Rn] $5f^{14}$ $6d^0$ $7s^2$
Lowrentium	103	Lr	[Rn] $5f^{14}$ $7s^2$ $7p^1$