## Physical properties of group 14 elements

Some of the physical properties of the group 14 elements are listed below

Property	Carbon	Silicon	Germanium	Tin	Lead
Physical state at 293 K	Solid	Solid	Solid	Solid	Solid
Atomic Number	6	14	32	50	82
Isotopes	<sup>12</sup> C, <sup>13</sup> C, <sup>14</sup> C	<sup>28</sup> Si, <sup>30</sup> Si	<sup>73</sup> Ge, <sup>74</sup> Ge	<sup>120</sup> Sn	<sup>208</sup> Pb
Atomic Mass (g.mol <sup>-1</sup> at 293 K)	12.01	28.09	72.63	118.71	207.2
Electronic configuration	[He]2s <sup>2</sup> 2p <sup>2</sup>	[Ne]3s <sup>2</sup> 3p <sup>2</sup>	$[Ar]3d^{10} 4s^2 4p^2$	[Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>2</sup>	[Xe] 4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>2</sup>
Atomic radius (Å)	1.70	2.10	2.11	2.17	2.02
Density (g.cm <sup>-3</sup> at 293 K)	3.51	2.33	5.32	7.29	11.30
Melting point (K)	Sublimes at	1687	1211	505	601
Boiling point (K)	4098	3538	3106	2859	2022